

New Triterpenoid Alkaloid Cholinesterase Inhibitors from *Buxus hyrcana*

M. Iqbal Choudhary,^{*,†} Salma Shahnaz,[†] Shehnaz Parveen,[†] Asaad Khalid,[†] S. Abdul Majeed Ayatollahi,[‡] Atta-ur-Rahman,^{*,†} and Masood Parvez[§]

HEJ Research Institute of Chemistry, University of Karachi, International Center for Chemical Sciences, Karachi-75270, Pakistan, School of Pharmacy, Shaheed Beheshti University of Medical Sciences, Tehran, Iran, and Department of Chemistry, University of Calgary, Calgary, Alberta, Canada T2N 1N4

Received September 13, 2002

Three new triterpenoid alkaloids, (+)-*N*-benzoylbuxahyrcanine [(20*S*)-3β-benzoylamino-20-dimethylaminobux-9(11)-ene-10α-ol] (**1**), (+)-*N*-tigloylbuxahyrcanine [(20*S*)-20-(dimethylamino)-3β-(2'-methyl-2'-butenoylamino)bux-9(11)-en-10α-ol] (**2**), and (+)-*N*-isobutyroylbuxahyrcanine [(20*S*)-20-(dimethylamino)-3β-(2'-methylpropanoyl)bux-9(11)-en-10α-ol] (**3**), have been isolated from the leaf extracts of *Buxus hyrcana* collected in Iran. Their structures were determined using spectroscopic methods. The structures of compounds **1** and **2** were unambiguously confirmed by single-crystal X-ray diffraction techniques. Compounds **1–3** were evaluated for their acetylcholinesterase and butyrylcholinesterase inhibitory activities, and compound **2** was found to be active against both enzymes.

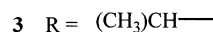
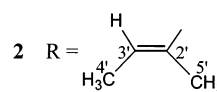
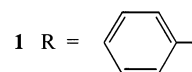
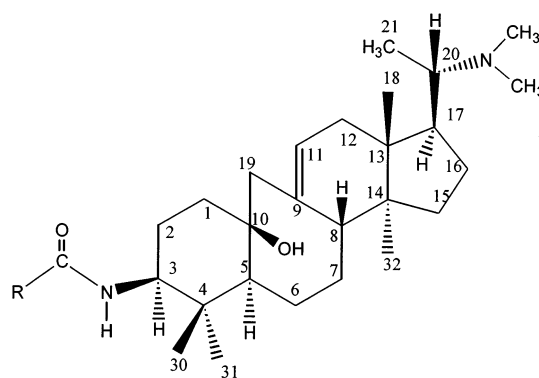
The *Buxus* plants are a rich source of triterpenoid alkaloids. Previous phytochemical studies on *Buxus* species have resulted in the isolation of more than 200 such compounds.¹ In the indigenous system of medicine, extracts of genus *Buxus* have been used for the treatment of various disorders such as malaria, rheumatism, and skin infections.² Continuing our studies on alkaloids of various *Buxus* species, work was started on the leaves of *Buxus hyrcana* Pojark. (Buxaceae). Seven compounds have been previously reported from this species.^{3–6} Continuing our studies on *B. hyrcana*, we report here the isolation, structure elucidation, and cholinesterase inhibitory activities of three new alkaloids, (+)-*N*-benzoylbuxahyrcanine (**1**), (+)-*N*-tigloylbuxahyrcanine (**2**), and (+)-*N*-isobutyroylbuxahyrcanine (**3**). The structures of these compounds were deduced from their spectral and crystallographic data.

Results and Discussion

The crude alkaloids were isolated from the methanolic extract of the leaves of *B. hyrcana* by extraction at different pH values. The fractions obtained at pH 7.0 were combined and evaporated to a gum, which was then subjected to column chromatography to yield alkaloids **1–3**.

(+)-*N*-Benzoylbuxahyrcanine (**1**) was isolated as a colorless amorphous material, which exhibited a peak at *m/z* 506 (HREIMS *m/z* 506.3812, C₃₃H₅₀N₂O₂, calcd 506.3872). The base peak at *m/z* 72.0814 (C₄H₁₀N, calcd 72.0813) indicated the presence of an *N,N*-dimethylamino group at C-20.⁷ Another peak at *m/z* 105.0380 (C₇H₅O, calcd 105.0340) was due to a benzoyl moiety. The UV spectrum showed absorption at 223 nm, characteristic of a secondary benzamide chromophore.⁸ The IR spectrum displayed intense bands at 3618 (OH), 3409 (NH, amide), 1660 (C=O, amide), and 1602 (C=C) cm⁻¹.

The ¹H NMR (CDCl₃) spectrum of **1** displayed four 3H singlets at δ 0.66, 0.96, 0.98, and 0.72 for the C-18, C-30, C-31, and C-32 tertiary methyl groups, respectively, while the secondary methyl resonated as a doublet at δ 0.82 (*J*_{20,21} = 6.5 Hz). A 6H broad singlet at δ 2.17 was assigned to the *N*₆-dimethyl protons. In turn, the proton resonating at



δ 3.91 (ddd, *J*_{3α,2β} = 12.5 Hz, *J*_{3α,NH} = 9.5 Hz, *J*_{3α,2α} = 3.5 Hz) was assigned to the C-3 methine proton, while the amide NH resonated as a doublet at δ 6.05 (*J*_{NH,3α} = 9.5 Hz). A doublet at δ 2.22 (*J*_{19α,19β} = 15.7 Hz) was due to H-19α, while H-19β appeared as a multiplet at δ 2.58. Finally, a downfield multiplet at δ 5.30 was ascribed to the C-11 olefinic proton, while the aromatic protons of the benzoyl moiety appeared as three sets of multiplets centered at δ 7.39, 7.45, and 7.72.

The ¹³C NMR spectra of **1** showed resonances for all 33 carbons including seven CH₃, eight CH₂, eleven CH, and seven quaternary carbons. The olefinic C-11 resonance appeared at δ 124.9, while its adjacent quaternary carbon resonated at δ 136.5. The C-3 signal was observed at δ 56.9, and its downfield shift suggested that a benzamide chromophore may be attached to this carbon. A quaternary carbon resonating at δ 73.0 suggested that **1** contains a tertiary OH group, whose probable position is C-10. Com-

* To whom correspondence should be addressed. Tel: +92-21-4990007. Fax: +92-21-4963124, -4963373. E-mail: zainraa@digicom.net.pk.

[†] HEJ Research Institute of Chemistry.

[‡] Shaheed Beheshti University of Medical Sciences.

[§] University of Calgary.

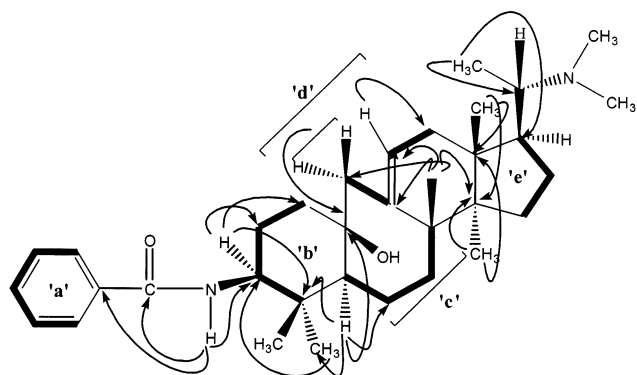


Figure 1. Important HMBC correlations and ^1H spin systems (bold lines) in **1**.

Table 1. ^{13}C and ^1H NMR Chemical Shift Assignments of **1**

carbon	$\delta^{13}\text{C}$	$\delta^1\text{H}$ (J, Hz)
1	26.9	1.45 m
2	25.4	1.65 m, 1.9 m
3	56.9	3.9 ddd (12.5, 9.5, 3.5)
4	38.9	
5	56.1	1.2 dd (12.5, 4.0)
6	27.1	1.52 m
7	28.2	1.3 m, 1.8 m
8	49.3	2.1 m
9	136.5	
10	73.00	
11	124.9	5.3 m
12	38.1	1.9 m
13	43.1	
14	49.7	
15	32.9	1.37 m
16	42.0	1.55 m, 1.65 m
17	49.2	1.8 m
18	16.0	0.66 s
19	52.7	2.22 d (15.9), 2.52 m
20	61.6	2.49 m
21	9.6	0.82 d (6.5)
30	16.2	0.96 s
31	27.4	0.98 s
32	16.2	0.72 s
N(CH ₃) ₂	39.9	2.17 s
1'	167.9	
2'	135.3	
3'	126.8	7.76 m
4'	128.5	7.40 m
5'	131.2	7.45 m
6'	128.5	7.40 m
7'	126.8	7.76 m

plete ^{13}C NMR chemical shift assignments and ^1H – ^{13}C direct one-bond connectivities of all protonated carbon atoms of **1**, as determined by the HMQC spectrum, are presented in Table 1.

The COSY-45° and HOHAHA (20, 60, and 100 ms) NMR spectra of **1** revealed the presence of five major spin systems (Figure 1). The cross-peaks in the HMBC spectrum were used to join the individual spin systems to obtain structure **1**. The NH (δ 6.05) and H-3 (δ 3.90) showed HMBC cross-peaks with the carbonyl carbon (δ 167.9), showing that the benzamide group is attached to C-3. H-3 (δ 3.90) and H-5 (δ 1.2) showed long-range correlations with C-4 (δ 38.9), while H-5 also showed HMBC interactions with C-10 (δ 73.0). This information suggested that spin systems b and e are connected to each other through C-4. The H-19 α and β signals (δ 2.22 and 2.58) showed long-range interactions with the C-10 (δ 73.0) and C-9 (δ 136.5) resonances, thereby indicating the presence of the OH group at C-10. Important HMBC interactions are shown in Figure 1.

A subsequent single-crystal X-ray diffraction analysis (Figure 2) unambiguously confirmed the structure **1** for this compound. The structure was solved by the direct methods⁹ and expanded by Fourier techniques.¹⁰ The figures were plotted with the aid of ORTEPII.¹¹ Most of the non-hydrogen atoms were refined anisotropically. A few atoms were found to be disordered. The tertiary OH group present at C-10 was found to be β oriented, showing that the A–B ring junction was trans. Phenyl rings were constrained as regular hexagons with C–C 1.39 Å and C–C–C 120°. Hydrogen atoms were included at geometrically idealized positions and were not refined. The refinement converged with $R_1 = 0.068$ and $wR_2 = 0.177$ with a goodness of fit of 1.02 for 6695 reflections with $F_o > 2\sigma(F_o)$. The largest peak on the final difference map was 0.26 e/Å³. The bond lengths and bond angles are within the usual ranges. Figure 2 shows a computer-generated perspective of the final X-ray model of **1**. The NOESY spectrum supported the stereochemistry determined by X-ray, in which cross-peaks were observed between H-19 α (δ 2.2), H-5 α (δ 1.2), and H₃-32 α (δ 0.72), which are only possible when ring B is disposed in *equatorial* position and the OH group is in *axial* position.

N-Tigloylbuxahyrcanine (**2**) (C₃₁H₅₂N₂O₂, m/z 484.3891, calcd 484.4028) was quite similar to **1** except that instead of a peak at m/z 105, an ion was observed at m/z 83 (composition C₅H₇O), which indicated the presence of a tigloyl group instead of a benzoyl group at C-3. The UV spectrum showed only terminal absorption at 206 nm, indicating the lack of any conjugated chromophore in the molecule. The ^1H and ^{13}C NMR spectra of **2** were similar to those of **1** except for signals from the tigloyl moiety which resonated at δ 168.9 (C-1'), 132.5 (C-2'), 129.0 (C-3'), 13.8 (C-4'), and 15.9 (C-5') and the absence of signals for the benzoyl substituent. In the NOESY spectrum, the H-3' signal showed interaction only with the C-4' methyl signal. The C-4' and C-5' methyls showed interactions with each other, thereby suggesting that the two methyls are in close vicinity to each other. The structure of **2** was then determined by single-crystal X-ray diffraction, which showed that 10-OH was β oriented. Figure 2 shows a computer-generated perspective of the final X-ray model of **2**. Therefore, the structure of compound **2** was determined as *N*-tigloylbuxahyrcanine.

N-Isobutyroylbuxahyrcanine (**3**) gave a M⁺ at m/z 472.4012 (C₃₀H₅₂N₂O₂, calcd 472.4028) in the HREIMS. The UV spectrum of **3** showed end absorption indicating the absence of any chromophore. The ^1H NMR spectrum of **3** showed a distinct resemblance with those of **1** and **2** except that it showed the presence of an isobutyroyl group (δ 1.13 and 1.15, d, each 3H, $J = 5.1$ Hz, C-3' and C-4' methyls, δ 2.3, m, 1H, C-2', compared with literature).¹² Several 2D ^1H NMR experiments (COSY-45°, HOHAHA, HMBC, and HMQC) further supported the chemical shift assignments for various protons, and in this way structure **3** was determined as *N*-isobutyroylbuxahyrcanine.

Extracts of several plants of the family Buxaceae have exhibited antiacetylcholinesterase properties.¹³ Furthermore, buxaminol E, a steroidal alkaloid isolated from *B. sempervirens*, has also been reported as a good inhibitor of the enzyme.¹⁴ We have previously reported the antiacetylcholinesterase activity of some *Buxus* alkaloids.⁶ Inhibition of acetylcholinesterase (AChE) is considered to be a promising approach for the treatment of Alzheimer's (AD) and related neurodegenerative diseases.^{15,16} The role of butyrylcholinesterase (BChE) in normal aging and brain diseases is still unclear, but recently BChE has been found in significantly higher quantities in Alzheimer's plaques

(CH₃)₂CO–C₆H₁₄–Et₂NH (2:7.9:0.1) were evaporated, and the solid mass thus obtained was repeatedly washed with acetone to yield *N*-isobutyrylbuxahyrcanine (**3**).

N-Benzoylbuxahyrcanine (1): colorless amorphous powder; 91.8 mg (% yield 3.22×10^{-4}) (CHCl₃); crystallized from MeOH; mp 239.7 °C; [α]_D²⁵ +15° (c 0.136, CHCl₃); UV λ_{max} (log ε) (MeOH) 223 (4.20) nm; IR (CHCl₃) ν_{max} 3648 (OH), 3409 (NH), 1660 (C=O), 1602 (C=C) cm⁻¹; ¹H NMR (CDCl₃, 500 MHz), see Table 1; ¹³C NMR (CDCl₃, 125 MHz), see Table 1; EIMS *m/z* 506 (0.1), 491 (0.2), 105 (5), 72 (100); HREIMS *m/z* 506.3812 (calcd for C₃₃H₅₀N₂O₂, 506.3872).

N-Tigloylbuxahyrcanine (2): colorless amorphous powder; 150 mg (% yield 5.26×10^{-4}) (CHCl₃); crystallized from MeOH–C₆H₆ (1:1); mp 245.5 °C; [α]_D²⁵ +62° (c 0.096, CHCl₃); UV λ_{max} (log ε) (MeOH) 206 (4.33) nm; IR (CHCl₃) ν_{max} 3547 (OH), 3452 (NH), 2839 (CH, aliphatic), 1640 (C=O, amide), 1622 (C=C) cm⁻¹; EIMS *m/z* 484 [M]⁺ (8), 469 (6), 83.0 (38), 72 (100); HREIMS *m/z* 484.3891 (calcd for C₃₁H₅₀N₂O₂, 484.4028).

N-Isobutyrylbuxahyrcanine (3): colorless amorphous powder; 169 mg (% yield 3.22×10^{-4}) (CHCl₃); mp 234–235 °C; [α]_D²⁵ +14° (c 0.124, CHCl₃); UV λ_{max} (log ε) (MeOH) 205.6 (4.33) nm; IR (CHCl₃) ν_{max} 3408 (OH), 2838 (CH, aliphatic), 1660 (C=O, amide) cm⁻¹; EIMS *m/z* 472 (8), 457 (6), 73 (50), 72 (100); HREIMS *m/z* 472.4012 (calcd for C₃₀H₅₂N₂O₂, 472.4028).

Crystal Data for 1: C₃₃H₅₀N₂O₂, mol wt = 506, monoclinic, space group *P*2₁, *a* = 14.2824(4) Å, *b* = 10.7080(3) Å, *c* = 20.2068(7) Å, *V* = 3037.21(16) Å³, *Z* = 4, *D*_c = 1.108 Mg/m³, *F*(000) = 1112, μ(Mo Kα) = 0.068 mm⁻¹. A single crystal of approximate dimensions 0.22 × 0.22 × 0.08 mm³ was used for all measurements. The intensity data of all unique reflections within the θ range 3.2–27.4° were collected at 295(2) K in a Nonius Kappa CCD diffractometer, using Mo Kα radiation on a graphite monochromator. A total of 7186 reflections were recorded, of which 6695 reflections with *F*_o [*I* > 2σ(*I*)] were taken into account for structure solution and refinements. The intensities were corrected for Lorentz and polarization factors and for absorption using the multiscan method.

Crystal Data for 2: C₃₁H₅₂N₂O₂, mol wt = 484, monoclinic, space group *P*2₁, *a* = 14.0960(3) Å, *b* = 10.7512(2) Å, *c* = 19.6728(5) Å, *V* = 2908.32(11) Å³, *Z* = 4, *D*_c = 1.107 Mg/m³, *F*(000) = 1072, μ(Mo Kα) = 0.068 mm⁻¹. A single crystal of approximate dimensions 0.25 × 0.22 × 0.08 mm³ was used for all measurements. The intensity data of all unique reflections within the θ range 3.7–27.5° were collected at 175(2) K in a Nonius Kappa CCD diffractometer, using Mo Kα radiation on a graphite monochromator. A total of 6901 reflections were recorded, of which 5156 reflections with *F*_o [*I* > 2σ(*I*)] were taken into account for structure solution and refinements. The intensities were corrected for Lorentz and polarization factors and for absorption using the multiscan method. Crystallographic data of **1** and **2**, including atomic coordinates, have

been deposited with the Cambridge Crystallographic Data Centre. Copies of the data can be obtained free of charge, on application to the director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK [fax: 44-(0) 1223-306033 or e-mail: deposit@ccdc.cam.ac.uk].

Cholinesterase Inhibition Assay. Acetylcholinesterase and butyrylcholinesterase inhibition were determined spectrophotometrically, using acetylthiocholine and butyrylthiocholine as substrates, respectively, by modifying the method of Ellman.¹⁸ The protocol has already been described in this journal.¹⁹

Acknowledgment. This work was supported by a research grant to M.I.C. from the Ministry of Science and Technology, Government of Pakistan, under the Pak-Kazakh scientific cooperation program.

Supporting Information Available: This material is available free of charge via the Internet at <http://pubs.acs.org>.

References and Notes

- Atta-ur-Rahman; Choudhary M. I. In *The Alkaloids*; Cordell, G. A., Ed.; Academic Press: San Diego, 1998; Vol. 50, Chapter 2, pp 61–108.
- Cordell, G. A. *Introduction to Alkaloids—A Biogenetic Approach*; Wiley-Interscience: New York, 1981; p 907.
- Aliev, A. M.; Orazmuradov, G. M. *Khim. Prir. Soedin.* **1974**, *10*, 409.
- Kurakina, I. O.; Tolkachev, O. N.; Pakali, D. A. *Khim. Prir. Soedin* **1974**, *10*, 814.
- Aliev, A. M.; Orazmuradov, G. M. *Khim. Prir. Soedin.* **1977**, *13*, 582.
- Atta-ur-Rahman; Parveen, S.; Khalid, A.; Farooq, A.; Ayatollahi, S. A. M.; Choudhary, M. I. *Heterocycles* **1998**, *49*, 481–488.
- Atta-ur-Rahman; Choudhary, M. I. In *Studies in Natural Product Chemistry*; Atta-ur-Rahman, Ed.; Elsevier Science Publishers: Amsterdam, 1988; Vol. 2, pp 176–204.
- Choudhary, M. I.; Atta-ur-Rahman; Freyer, A. J.; Shamma, M. *J. Nat. Prod.* **1987**, *50*, 84–88.
- Altomare, A.; Cascarano, M.; Giacobozzo, C.; Guagliardi, A. *SIR92. J. Appl. Crystallogr.* **1993**, *26*, 343.
- Beurskens, P. T.; Admiraal, G.; Beurskens, G.; Bosman, W. P.; de Gelder, R.; Israel, R.; Smits, J. M. M. The *DIRDIF-94* program system; Technical Report of the Crystallography Laboratory; University of Nijmegen: The Netherlands, 1994.
- Johnson, C. K. *ORTEPII*; Report ORNL-5138; Oak Ridge National Laboratory, TN, 1976.
- Wu, J. B.; Chun, Y. T.; Ebizuka, Y.; Sanakawa, U. *Chem. Pharm. Bull.* **1991**, *39* (12), 3272–3275.
- Kiamuddin, M.; Hye, H. K. M. A. *Pak. J. Sci. Ind. Res.* **1970**, *13*, 59–62.
- Kvaltinova, Z.; Lukovic, L.; Machova, J.; Fatranska, M. *Pharmacology* **1991**, *43*, 20–25.
- Quinn, D. M. *Chem. Rev.* **1987**, *87*, 955–979.
- Nochi, S.; Asakawa, N.; Sato, T. *Biol. Pharm. Bull.* **1995**, *18*, 1145–1147.
- Yu, Q.-S.; Holloway, H. W.; Utsuki, T.; Brossi, A.; Greig, N. H. *J. Med. Chem.* **1999**, *42*, 1855–1861.
- Ellman, G. L.; Courtney, D.; Andres, V.; Featherston, R. M. *Biochem. Pharmacol.* **1961**, *7*, 88–95.
- Atta-ur-Rahman; Fatima, N.; Akhtar, F.; Choudhary, M. I.; Khalid, A. *J. Nat. Prod.* **2000**, *63*, 1393–1395.

NP0204460